

**Causes of Respiratory Acidosis (high H<sup>+</sup>, high CO<sub>2</sub>)****Hypoventilation**

- **Obstructive pulmonary disease:** COPD, asthma
- **Upper airway obstruction**
- **Mechanical obstruction to lung expansion:** pneumothorax, pleural effusion, chest wall deformities
- **Neuromuscular disease:** Guillain-Barre syndrome, myasthenia gravis
- **CNS depression:** sedatives, stroke, intracranial bleed

**Causes of Respiratory Alkalosis (low H<sup>+</sup>, low CO<sub>2</sub>)****Stimulated respiratory drive**

- **CNS:** stroke, intracranial bleed, psychogenic
- **Hypermetabolic:** thyrotoxicosis, pregnancy, fever, delirium tremens
- **Hyperthermia**
- **Iatrogenic mechanical ventilation**

**Hypoxia-induced****Compensation for metabolic acidosis****Causes of Metabolic alkalosis (low H<sup>+</sup>, high HCO<sub>3</sub><sup>-</sup>)****Loss of acid**

- **Loss from GI tract:** vomiting, ileostomy
- **Loss from renal tract:** hyperaldosteronism, diuretics

**Gain of alkali**

- **Bicarbonate infusion**
- **Excessive antacid/laxative consumption**
- **Massive blood transfusion** (citrate in PRC converted to bicarbonate)

## Causes of Metabolic Acidosis (high H<sup>+</sup>, low HCO<sub>3</sub><sup>-</sup>)

### Increased acid production

- **Tissue hypoxia:** cardiorespiratory depression or impaired oxygen carrying capacity (carbon monoxide poisoning, methaemoglobinemia)
- **Production of ketoacids:** DKA, alcoholic ketoacidosis and ethanol/isoniazid poisoning
- **Interference with ATP usage:** paracetamol, valproate, metformin, CO, cyanide

### Exogenous acids

- **Poisoning by acidic substances:** salicylates
- **Poisoning by substances with acidic metabolites:** methanol & ethylene glycol

### Loss of bicarbonate

- **Loss from GI tract:** diarrhoea, pancreatic fistula
- **Loss from the renal tract:** renal tubular acidosis

### Decreased acid elimination

- **Acute kidney injury**
- **Toxic metabolites causing renal impairment:** ethylene glycol

The **anion gap** can be calculated to help work out the cause of a metabolic acidosis. The anion gap is equal to the difference between the plasma concentrations of the measurable cations (positive ions: Na<sup>+</sup> and K<sup>+</sup>) and anions (negative ions: Cl<sup>-</sup> and HCO<sub>3</sub><sup>-</sup>):

$$(\text{Na} + \text{K}) - (\text{HCO}_3 + \text{Cl})$$

A **normal anion gap** is <18mmol/L. A high anion gap is associated with the addition of endogenous or exogenous acids which are paired with an unmeasured cation.

Metabolic acidosis with raised anion gap	Metabolic acidosis with normal anion gap
Methanol	GI HCO <sub>3</sub> <sup>-</sup> loss (eg diarrhoea)
Uraemia	Renal HCO <sub>3</sub> <sup>-</sup> loss (renal tubular acidosis)
Diabetic ketoacidosis	Renal failure
Paraldehyde	Hypoaldosteronism
Iron/isoniazid	Carbonic anhydrase inhibitors
Lactic acid	Chloride ingestion (eg magnesium chloride)
Ethanol/ethylene glycol	
Salicylates	